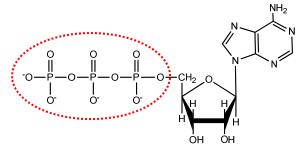
### ATP – adenosine triphosphate

- · Energy currency (ex: muscle contraction)
- Phosphoryl donor (ex: hexokinase reaction of glycolysis)
- · Pyrophosphoryl donor (ex: thiamine diphosphokinase reaction)
- · AMP donor (ex: amino acid activation for protein synthesis)
- · Nucleotide in RNA synthesis
- · Allosteric effector (ex: phosphofructokinase reaction of glycolysis)
- · (See Lehninger p23, 501-503, 506-509; 178, 212, 532)

## ADP – adenosine diphosphate

- · Phosphoryl donor (ex: adenylate kinase reaction)
- Phosphoryl acceptor (ex: pyruvate kinase reaction of glycolysis)
- · Allosteric effector (ex: isocitrate dehydrogenase reaction of the TCA cycle)
- · (See Lehninger p23; 510, 536-537, 538)



Adenosine triphosphate (ATP)

Adenosine diphosphate (ADP)

### AMP – adenosine monophosphate

- · Phosphoryl acceptor (ex: adenylate kinase reaction)
- · Allosteric effector (ex: glycogen phosphorylase reaction of glycogen catabolism)
- · (See Lehninger p 23, 273)

Adenosine monophosphate (AMP)

### CoA (or CoASH) - coenzyme A

- · Thioester energy currency (ex: succinyl-CoA synthetase reaction of the TCA cycle)
- · Acyl carrier via thioester formation (ex: as acetyl-CoA entering the TCA cycle)
- · Synthesized from pantothenic acid, essential vitamin B<sub>5</sub>
- (See Lehninger p 617; 505, 616-619, 622-623, 626)

Coenzyme A (CoA)

### FMN (redox forms FMN, FMNH•, or FMNH<sub>2</sub>) – flavin mononucleotide

- · Redox cofactor:
  - accepts/gives 2 electrons and 2 protons (= hydride ion + proton), (ex: Complex I reaction of ETC)
  - accepts/gives 1 electron and 1 proton (= hydrogen atom), (ex: Complex I reaction of ETC)
- · Tightly bound in enzyme active site
- · Varies in redox potential based on protein surroundings
- · Synthesized from riboflavin, essential vitamin B<sub>2</sub>
- · (See Lehninger p 519-521, 709, 714)

Flavin mononucleotide (FMN)

Flavin adenine dinonucleotide (FAD)

## FAD (redox forms FAD, FADH•, or FADH2) - flavin adenine dinucleotide

- Redox cofactor:
  - accepts/gives 2 electrons and 2 protons (= hydride ion + proton), (ex: succinate DH/Complex II reaction of the TCA cycle and ETC)
  - accepts/gives 1 electron and 1 proton (= hydrogen atom), (ex: succinate DH/Complex II reaction of the TCA cycle and ETC)
- · Tightly bound in enzyme active site
- · Varies in redox potential based on protein surroundings
- · Synthesized from riboflavin, essential vitamin B<sub>2</sub>
- · (See Lehninger p 519-521, 628, 709, 715)

 $NH_2$ 

-OH in NAD

Nicotinamide adenine dinucleotide phosphate

(NADP)

### NAD (redox forms NAD<sup>+</sup> or NADH) – nicotinamide adenine dinucleotide

- · Redox cofactor: accepts/gives 2 electrons and 1 proton (= hydride ion), (ex: GAPDH reaction of glycolysis)
- · Water soluble and freely diffusing
- · Used mainly in catabolic pathways
- $\cdot$  [NAD<sup>+</sup>] > [NADH] in cells
- · Synthesized from niacin or niacinamide (nicotinamide), essential vitamin B<sub>3</sub>
- · (See Lehninger p 516-519; 535-536, 616-619, 624-625, 628, 709)

$$\begin{array}{c|c} H & O \\ \hline \\ NH_2 \\ \hline \\ R \\ \end{array} \begin{array}{c} H^+ + 2e^- \\ \hline \\ R \\ \end{array} \begin{array}{c} H \\ \hline \\ NH_2 \\ \hline \\ R \\ \end{array}$$

NAD+ or NADP+

NADH or NADPH

# NADP (redox forms $NADP^+$ or NADPH) – nicotinamide adenine dinucleotide phosphate

- Redox cofactor: accepts/gives 2 electrons and 1 proton (= hydride ion), (ex: G6P DH reaction of the pentose phosphate pathway)
- · Water soluble and freely diffusing
- · Used mainly in anabolic pathways
- ·  $[NADP^+] < [NADPH]$  in cells
- · Synthesized from niacin or niacinamide (nicotinamide), essential vitamin B<sub>3</sub>
- · (See Lehninger p 516-519; 624)

## CoQ (or Q; redox forms Q, QH•, QH2) - coenzyme Q or ubiquinone

- · Redox cofactor: accepts/gives 1 electron and 1 proton (= hydrogen atom) up to 2 times, (ex: Q-cycling at Complex III)
- · Hydrophobic; freely diffusing in the membrane
- · Length of isoprene tail varies by species
- · (See Lehninger p 710, 715-717)

### **Biotin**

· Helps catalyze carboxylation (ex: pyruvate carboxylase reaction of gluconeogenesis)

· Covalently linked (via amide linkage) to enzyme at lysine side chain

· Essential vitamin B<sub>7</sub>

· (See Lehninger p554, 633-635)

## Lipoic acid (redox forms 'lipoic acid' and 'dihydrolipoic acid')

· Redox cofactor; accepts 2 electrons (plus 2 protons) at disulfide, yielding sulfhydryls (ex: PDH complex reaction in aerobic metabolism)

- · Covalently linked (via amide linkage) to enzyme at lysine sidechain; becomes lipoamide (or dihydrolipoamide)
- Catalyzes acyl transfer via thioester formation (ex: PDH complex reaction in aerobic metabolism)
- · (See Lehninger p 617, 635, 616-619)

### **TPP** – thiamine pyrophosphate

- · Catalyzes α-keto decarboxylation (ex: PDH complex reaction of aerobic metabolism)
- · Ylid proton is acidic, with pKa ~ 18 (pKa may differ in enzyme)
- · Bound tightly in enzyme active site
- · Synthesized from thiamine, essential vitamin  $B_1$
- · (See Lehninger p 550, 616-619)

Thiamine pyrophosphate (TPP)

$$R$$
 $R$ 
 $R$ 
 $R$ 
 $R$ 
 $R$ 
 $R$ 
 $R$ 
 $R$ 
 $R$ 

#### Heme

- · O<sub>2</sub>-binding cofactor
- · Redox cofactor: accepts/gives 1 electron (Fe<sup>3+</sup> + e<sup>-</sup> ↔ Fe<sup>2+</sup>), (ex: Complex IV reaction of ETC)
- · Bound tightly in protein, either non-covalently via hydrophobic interactions, or covalently
- · Varies in redox potential based on R-groups and protein surroundings
- · (See Lehninger p 154, 710-711, 716-718)