- 1. Which of the reactions of a metabolic pathway would you expect to be regulated? List all that apply:
 - a. the 1st committed step of the pathway
 - b. the last step of the pathway
 - c. a highly spontaneous reaction
 - d. a rate-limiting reaction
 - e. a reaction in which [products]/[reactants] is close to K_{eq}
 - f. a very fast reaction (relative to others in the pathway)
 - g. a reaction with a large, positive ΔE°
 - h. a reaction involving a phosphoryl transfer from phosphoenolpyruvate
 - i. a reaction that is used in the reverse pathway
- 2. Consider the structure of inorganic phosphate (P_i)
 - a. Does P_i act as a nucleophile or electrophile in biochemical reactions?
 - b. Briefly explain your answer for part **a** (15 words or fewer).
 - c. The hydrolysis of phosphorylated compounds is generally spontaneous:

 $R - O - PO_3^{2-} + H_2O = R - OH + P_i$ $\Delta G'^{\circ} = negative$

What property of P_i favors this reaction? Briefly explain (15 words or fewer).

- 3. List three factors that contribute to the large release of free energy when ATP is hydrolyzed.
- 4. What information would you need to determine whether a compound containing phosphate has a high or low phosphoryl transfer potential?
- 5. We discussed different redox energy currencies, such as NAD, NADP, and FAD. Why does the cell have so many different redox currencies? Why isn't one sufficient?
- 6. Which of the following catalytic mechanisms are likely to be important in the isomerization of glucose-6-phosphate to fructose-6-phosphate:
 - a. general acid catalysis
 - b. general base catalysis
 - c. covalent catalysis
 - d. electrostatic catalysis
 - e. metal-ion catalysis
- 7. Define 'isozyme.'
- 8. True or False? Glucosephosphate isomerase and triosephosphate isomerase are isozymes.
- 9. Does glucose or ATP act as a nucleophile in the reaction catalyzed by hexokinase? Briefly explain your answer (15 words or fewer).
- 10. Generalize your answer to question 9 by selecting the correct words to completing the following sentence:

In phosphoryl transfer reactions, the phosphoryl donor is the (nucleophile/electrophile) and the phosphoryl acceptor is the (nucleophile/electrophile).

- 11. Which one of the following sets most likely represents amino acids found in the active site of phosphoglucose isomerase (PGI)?
 - a. Arg, Ser, Met, Val
 - b. Phe, Val, Ile, Thr
 - c. Ser, Cys, His, Gln
 - d. Tyr, Thr, Trp, His
 - e. Gly, Ala, Cys, Pro
- 12. Which of the following catalytic mechanisms are used by aldolase?
 - a. general acid catalysis
 - b. general base catalysis
 - c. covalent catalysis
 - d. electrostatic catalysis
 - e. metal-ion catalysis
- 13. Many enzymes of glycolysis require metal ion cofactors. Why? How do these cofactors aid in catalysis?
- 14. The ATP made in glycolysis is said to be made via 'substrate-level phosphorylation.' Which enzymes of glycolysis are involved in substrate-level phosphorylation? What do these enzymes (and the reactions they catalyze) have in common?