

Summary of Alkene Chemistry

<u>Reaction</u>		<u>Conditions</u>	<u>Regiochemistry</u>	<u>Stereochemistry</u>	<u>Rearrangements</u>
Dehydration	- H ₂ O	Strong acid	Zaitsev's Rule		Yes
	- H ₂ O	POCl ₃ in pyridine	Zaitsev's Rule		No
Dehydrohalogenation	- HX	Strong base (RO ⁻ or R ₂ N ⁻)	Zaitsev's Rule	Anti-coplanar	No
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Hydrogenation	H ₂	Pt, Pd, Ni, etc.		Syn Addition	No
Hydration	H ₂ O	Aqueous acid	Markovnikov		Yes
	H ₂ O	1. Hg(OAc) ₂ / H ₂ O 2. NaBH ₄	Markovnikov		No
	H ₂ O	1. R ₂ BH; 2. H ₂ O ₂ , OH ⁻	Anti-Markovnikov	Syn Addition	No
Halogen Addition	Br ₂ , Cl ₂			Anti Addition	No
Halohydrin Addition	HOBr, HOCl	X ₂ and H ₂ O, or NBS or NCS in H ₂ O	Markovnikov	Anti Addition	No
Hydrogen Halide Addition	HCl, HBr		Markovnikov		Yes
	HBr	air or peroxides	Anti-Markovnikov		No
Epoxidation	"O"	Peroxyacid		Syn Addition	No
Hydroxylation	two -OHs	OsO ₄ / H ₂ O ₂ or KMnO ₄ / NaOH, cold		Syn Addition	No
Ozonolysis	"O ₂ "	1. O ₃ ; 2. Zn / H ₂ O or (CH ₃) ₂ S	(σ and π bond cleavage to give aldehydes and ketones)		