CHEMISTRY 130 BL QUIZ 2 (Wittig-Reaction) Fall 1998 (10-19-98)

Fill out the following information completely or you will loose 5 points. Please indicate who your TA is or in which section you are. It would be also nice if you could write legible. You have 10 minutes to complete the quiz. Good luck.

First letter of your last name:

Full Name:

UCLA I.D.#:

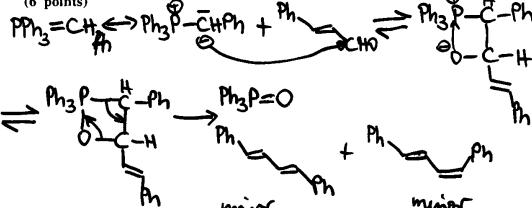
Section/TA: Zhe Joe

Thach

1. The methylene group in benzyltriphenylphosphonium chloride (Wittig salt) appears as a doublet at δ =5.5 ppm. Explain briefly. (2 points)

-doublet because of coupling wik 31P (I=1k) - chemical shift mainly due to positive charge on P

2. Outline the mechanism for the formation of the final product starting from the ylid and cinnamaldehyde. Show all products of the reaction and the correct stereochemistry! (6 points)



3. Which color does

a) the ylid

b) the major product

c) the minor product have? (3 points)

yellowish

4. What are the health hazards of benzylchloride and triphenylphosphin? (3 points)

Benzylchloride:lachrymator(kar-causing), skin irritant Triphenylphophin: toxic!

5. How can you distiguish between the major and the minor product by

a) IR spectroscopy (2 points)

trans, trans: coplending at ~ 960cm - 1 and 730cm - 1

b) H-NMR spectroscopy (2 points)

trans, trans: two signals for alkene H, Ja 16 Hz

transicis: four signals for alkenett, Je 1642, Je 842

c) C-NMR spectroscopy (2 points)

trans, trans: Symmetry, 6 Signals

trans, cis: no symmetry, 12 signals

d) physical properties (2 points)

t,t: mp=152° & different R_f values | Solubility in E+OH

t,c: oil at r.t. | R_f(t,c)>R_f(t,t)

6. Predict the products of the following reactions including the correct stereochemistry (6 points).

Ph₃P + PhCH₂Br
$$\frac{150 \text{ C}}{\text{Xylene}}$$
 Ph₃P - CH₂Ph NaOEt - Th₃P - CH₂Ph NaOEt + Na & CH₂Ph NaOEt + EtoH